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The Blue Bead Treasure Trove Workbook Answers - ICSE Solutions

These reactions are also called redox reactions (see p. 82). e.g. $\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{g})$ Neutralisation reactions During neutralisation reactions, a base reacts with an acid to produce a salt and water (see p. 77). e.g. $2\text{KOH}(\text{aq}) + \text{H}_2\text{SO}_4(\text{aq}) \rightarrow \text{K}_2\text{SO}_4(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$ Reversible reactions During a reversible reaction, the direction of the chemical change can be easily reversed.

Collins - Concise Revision Course For CSEC Chemistry | PDF ...

CHM130 pH and Buffer lab pH Measurements and Buffer Laboratory Introduction: pH is a measure of the acidity of an aqueous solution. It is related to the concentration of hydrogen ion, H^+ . The pH scale can tell if a liquid is more acid or more base,

pH Measurements and Buffer Laboratory Introduction

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